

Ultrafast spectroscopy of liquid solutions under vacuum

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Liquids under vacuum?

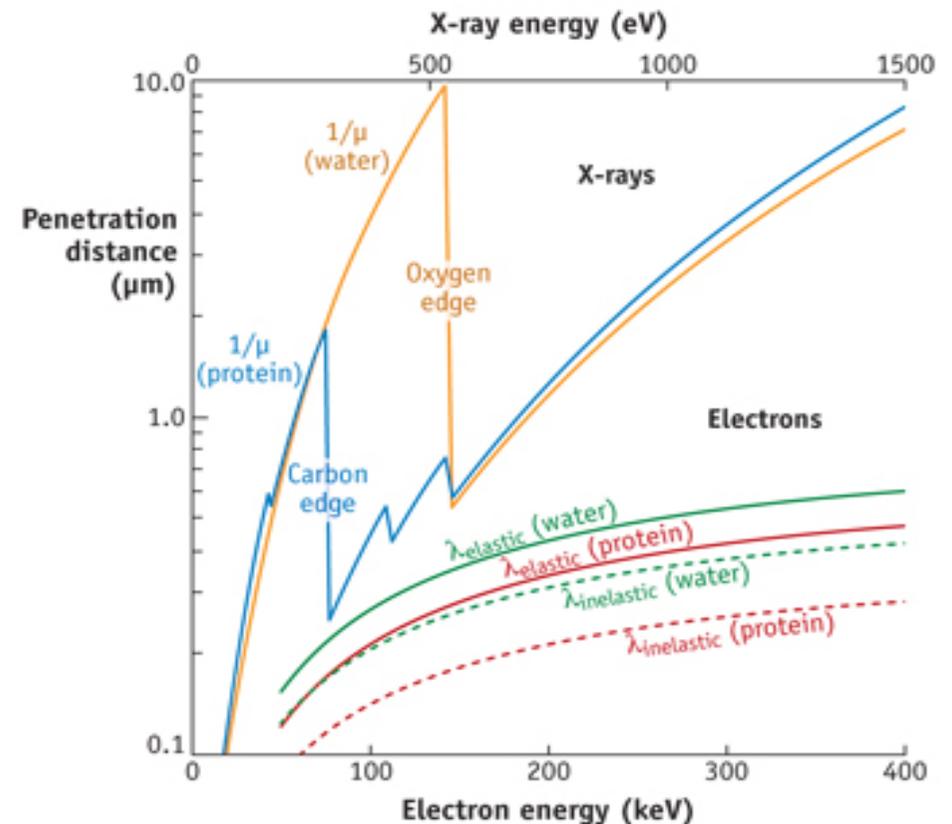
- Photoelectron spectroscopy of liquid solutions (ESCA)
- EUV to soft X-ray absorption of solutions
- Ultrahigh time-resolution spectroscopy (minimize GVD/GVM)

Spectroscopy in the EUV/soft X-ray range (< 1000 eV)

- Water window (282 to 530 eV)
- K-edges of low-Z elements : C, N, O, S, ... Interest for molecular solids, catalysis and chemisorption processes, biology, etc.
- L_{2,3}-edges of transition metals (2p_{½,3/2}- nd): interest for biology, coordination chemistry, catalysis, magnetism, etc.
- Sharp spectral features: $\Gamma_L = 0.5 \text{ eV} = \hbar / t = 6.58 \cdot 10^{-16} / t$

Challenges:

- Need for vacuum (problem with volatile samples)
- High fluxes: $\alpha_{\text{abs}} \sim Z^4$ (Low-Z elements)
- Solvent absorption (need for μm -thick samples)



Contents

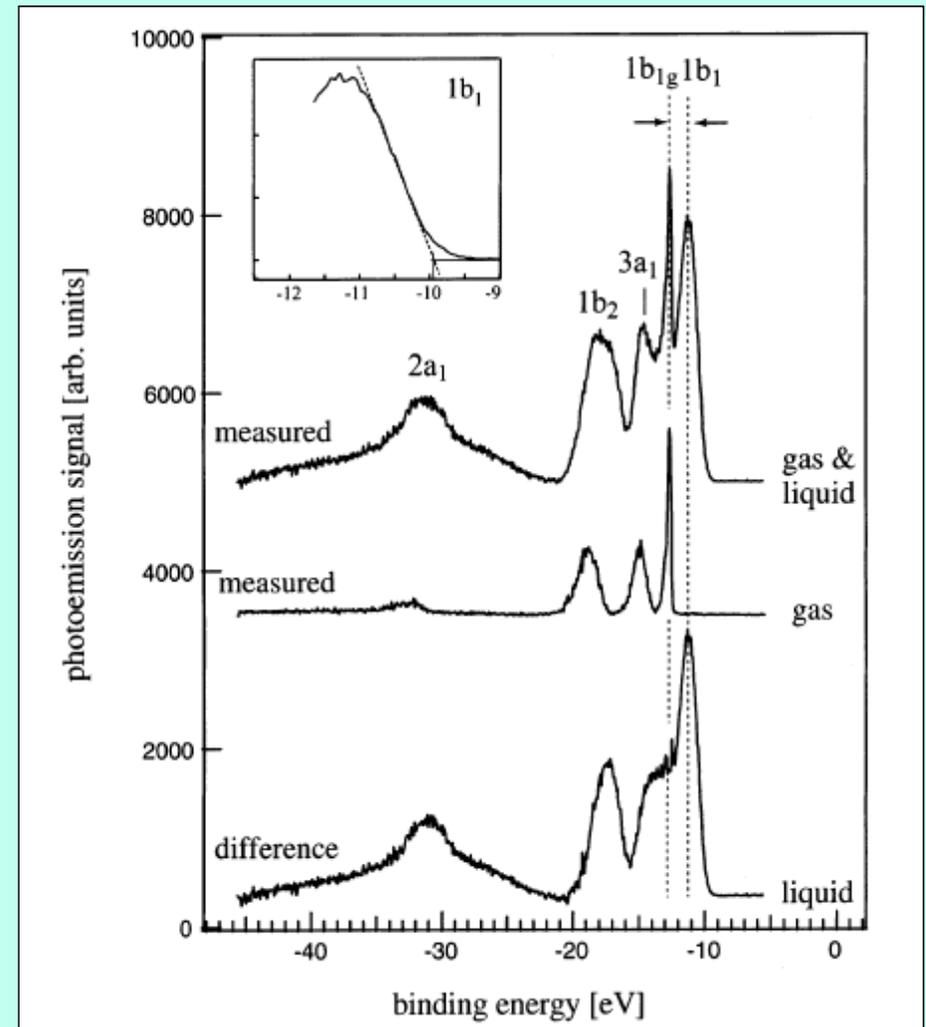
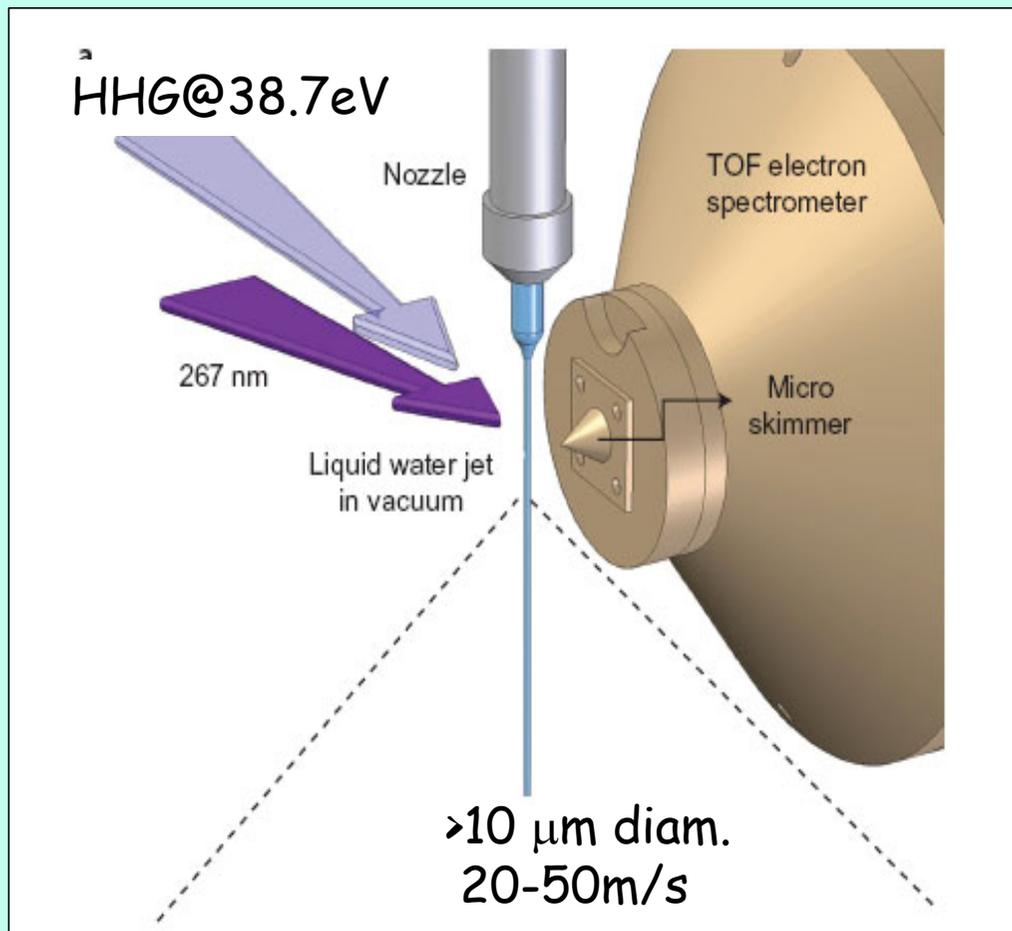
- I. Ultrafast photoelectron spectroscopy of charge transfer reactions in solutions (cylindrical jets)
- II. EUV/soft X-ray absorption studies of molecular systems (Flat jets)
- III. ultrahigh time resolution of liquid solutions (Flat jets)

Steady-state UPS/XPS in liquids

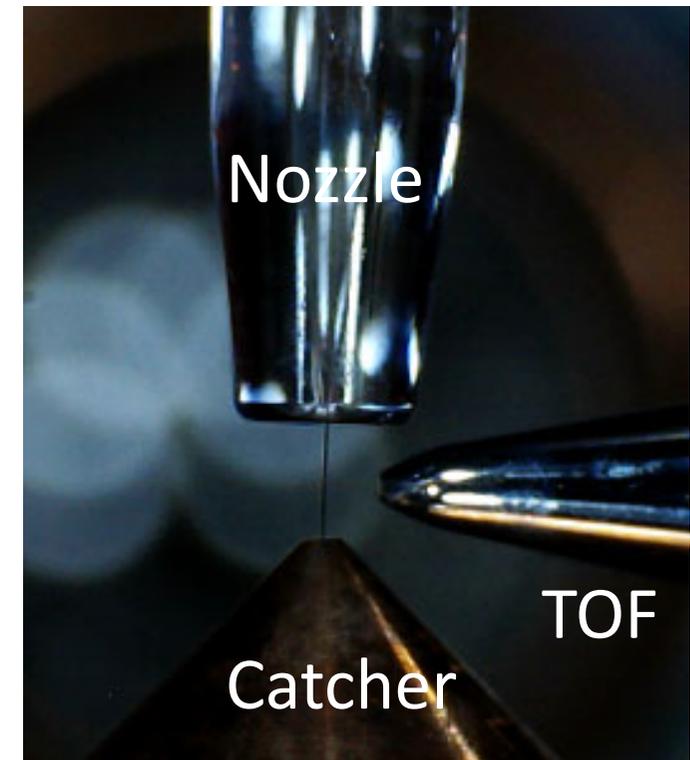
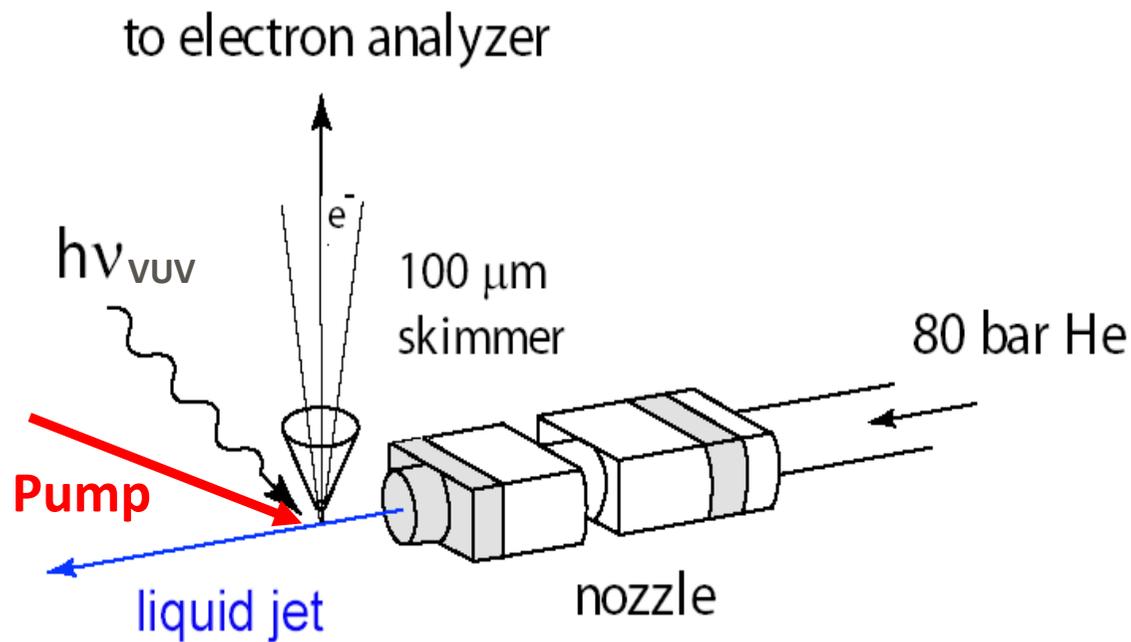
Cylindrical jets: typ. 50 μm diameter
Faubel et al. *J. Chem. Phys.* 1997 & *JPCA* (2004);
Winter and Faubel, *Chem. Rev.* (2006)

Ultrafast VUV-PES of liquids

Abel, Faubel et al. *Nat. Chem.* 2010 &
Appl. Phys. A 2009



Ultrafast (<20 fs) Vacuum ultraviolet photoelectron spectroscopy of solutions

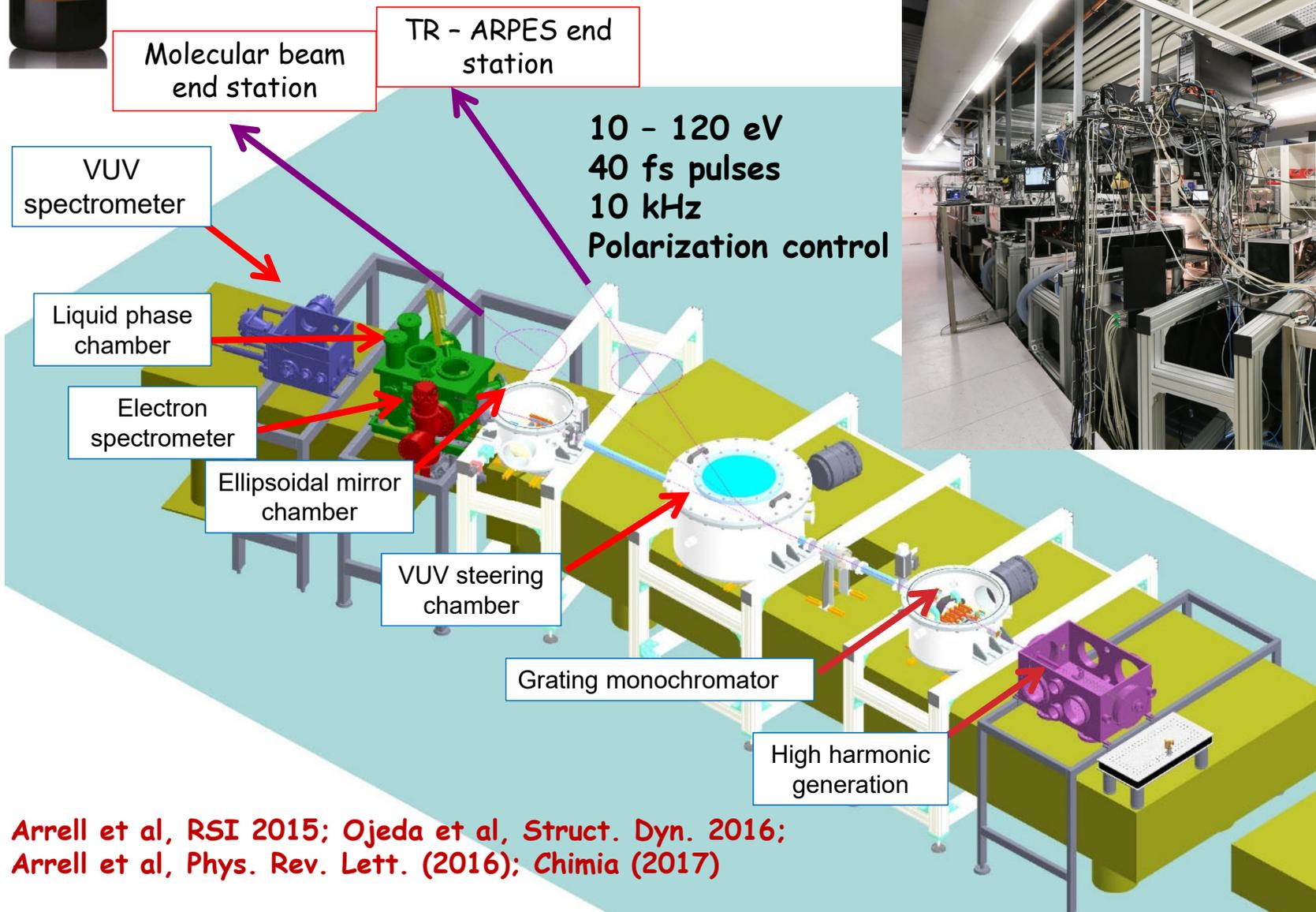


Arrell et al, RSI 2014

- Rapid ejection of liquid through $\sim 20 \mu\text{m}$ capillary creates stable region of 1-2 mm laminar flow
- Cryo pumping used to maintain $\sim 10^{-5}$ mbar conditions
- Probing by EUV pulses from the Harmonium HHG source



Extreme-UV femtosecond source:
Facility for photoelectron spectroscopy (ESCA) of liquid, gas and solid phases.
Complementary to X-ray studies at the SLS and XFELs

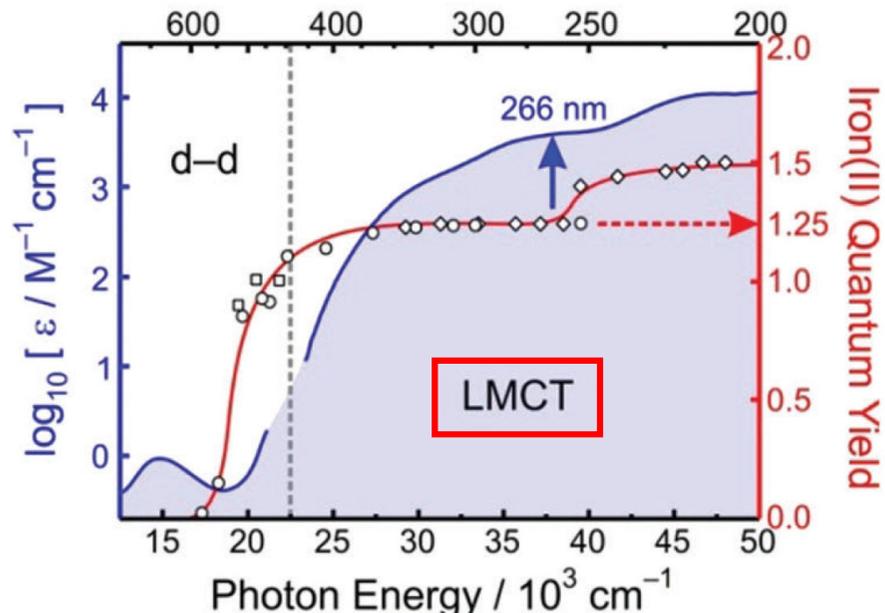
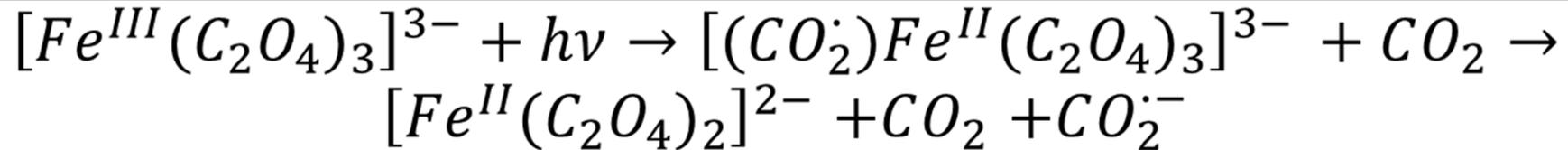
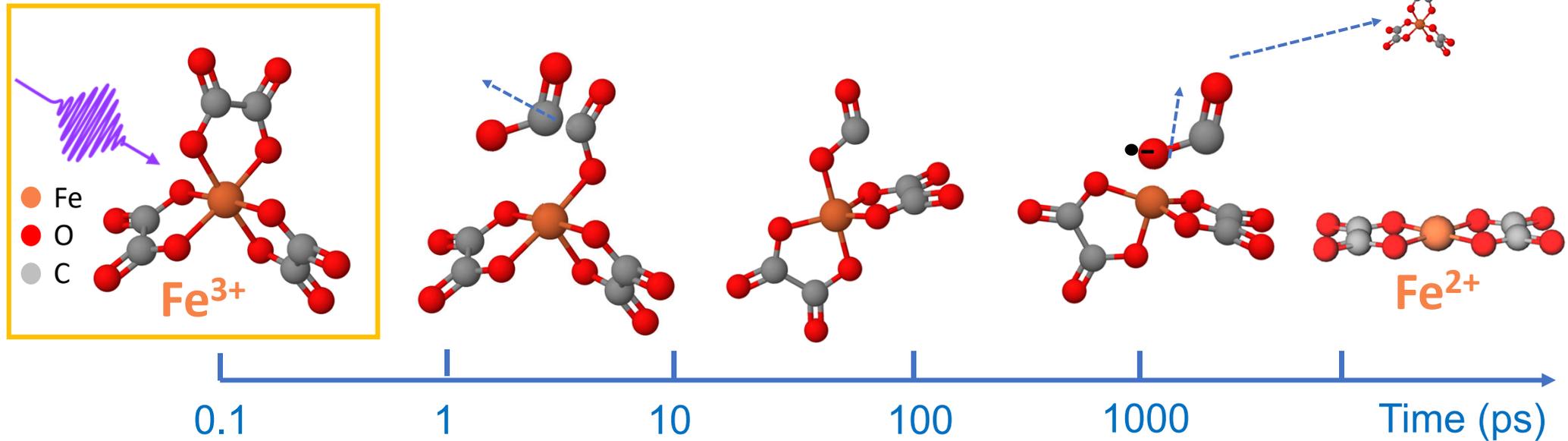


Funding:



Arrell et al, RSI 2015; Ojeda et al, Struct. Dyn. 2016;
Arrell et al, Phys. Rev. Lett. (2016); Chimia (2017)

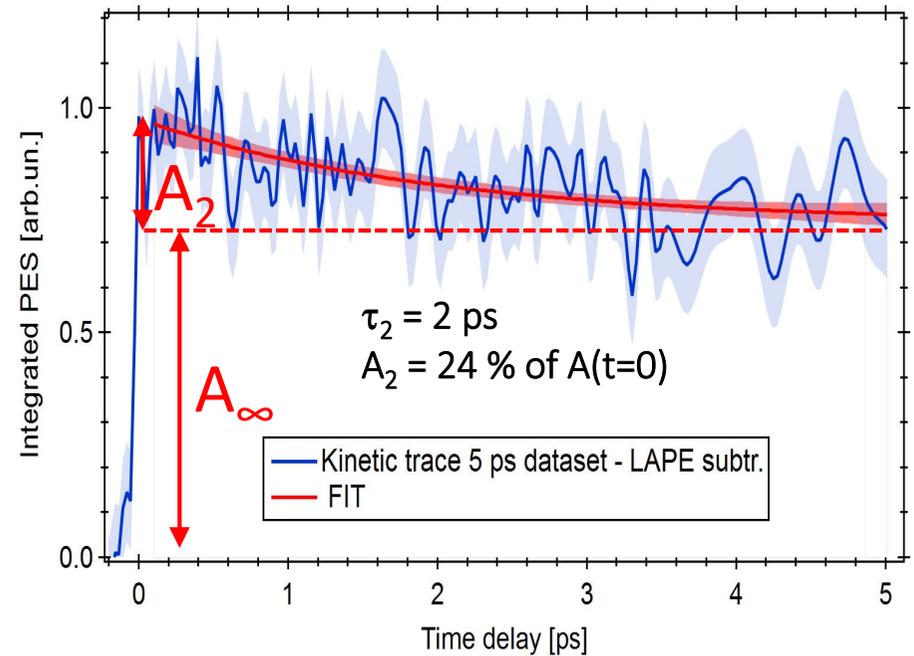
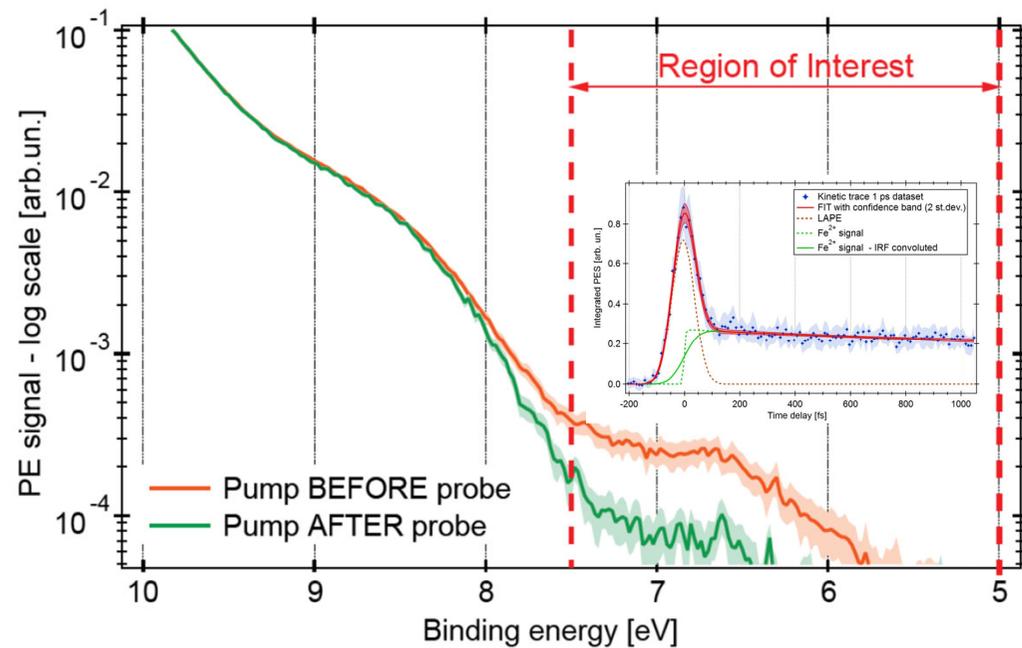
Ultrafast PES of a photochemical reaction (Ferrioxalate)



Parker & Hatchard, *J. Phys. Chem.* 1959; Rentzepis & Co, *Inorg. Chem.* 2008; Suzuki & Co, *Struct. Dyn.* 2015; Straub *et al.* *PCCP* 2018;

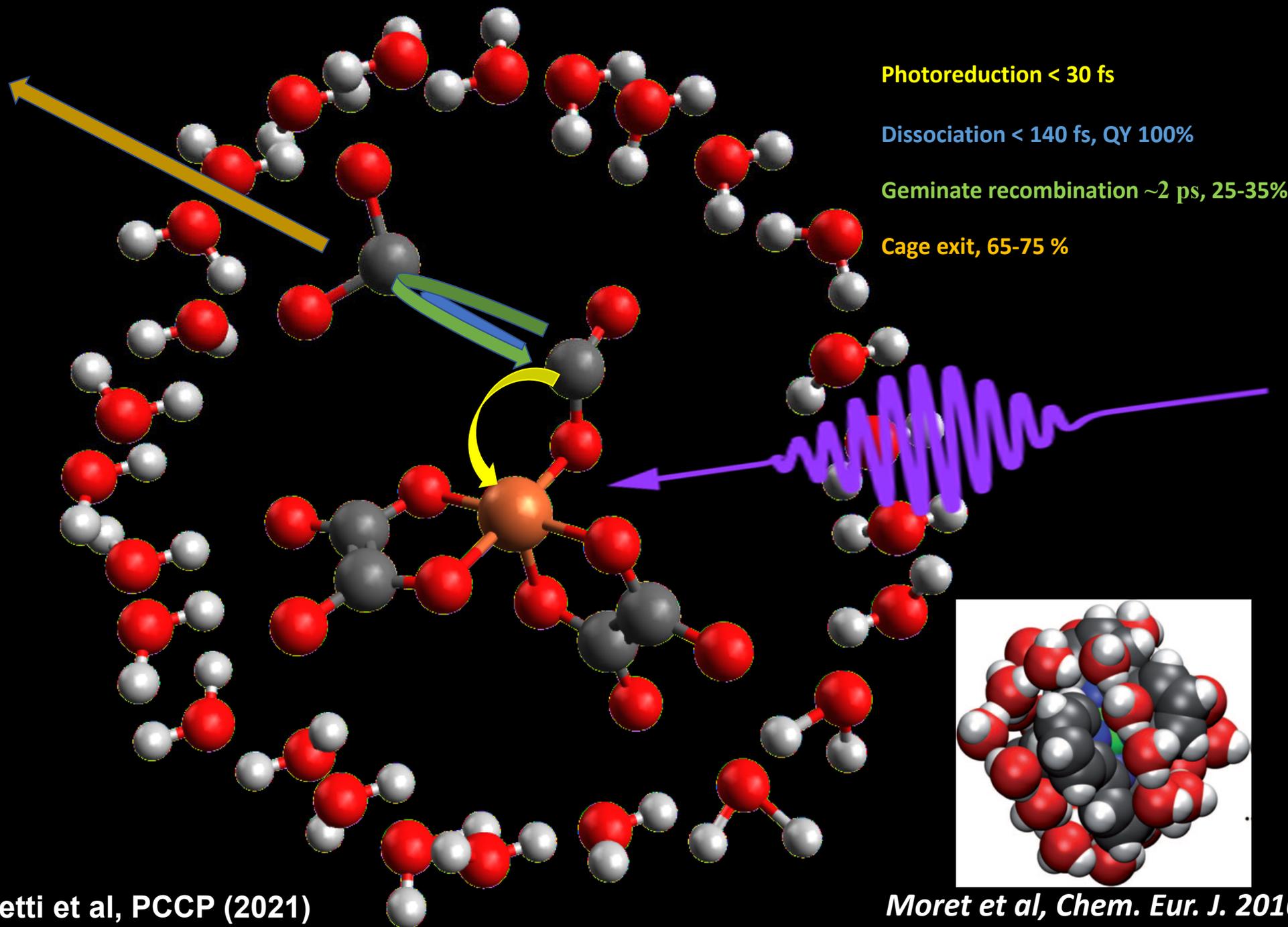
- How fast is the photoreduction?
- How fast is the first CO_2 dissociation?
- Does reduction trigger dissociation or vice-versa?
- What is the role of the solvent?

Aqueous Ferrioxalate

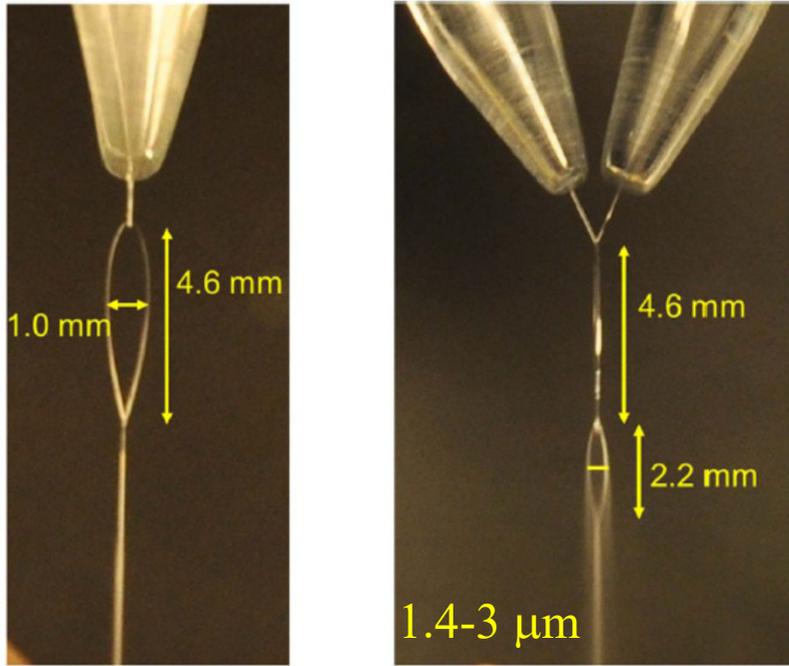


- 25% of ferrous species are lost in 2 ps
- IR TA: partial (25-35 %) recovery of depleted parent molecule population in ca. 2 ps. Attributed to intramolecular relaxation.
- X-ray TA: <140 fs CO₂ dissociation and 2-3 ps relaxation time, attributed to dissociation of the second fragment (the CO₂⁻ anion)

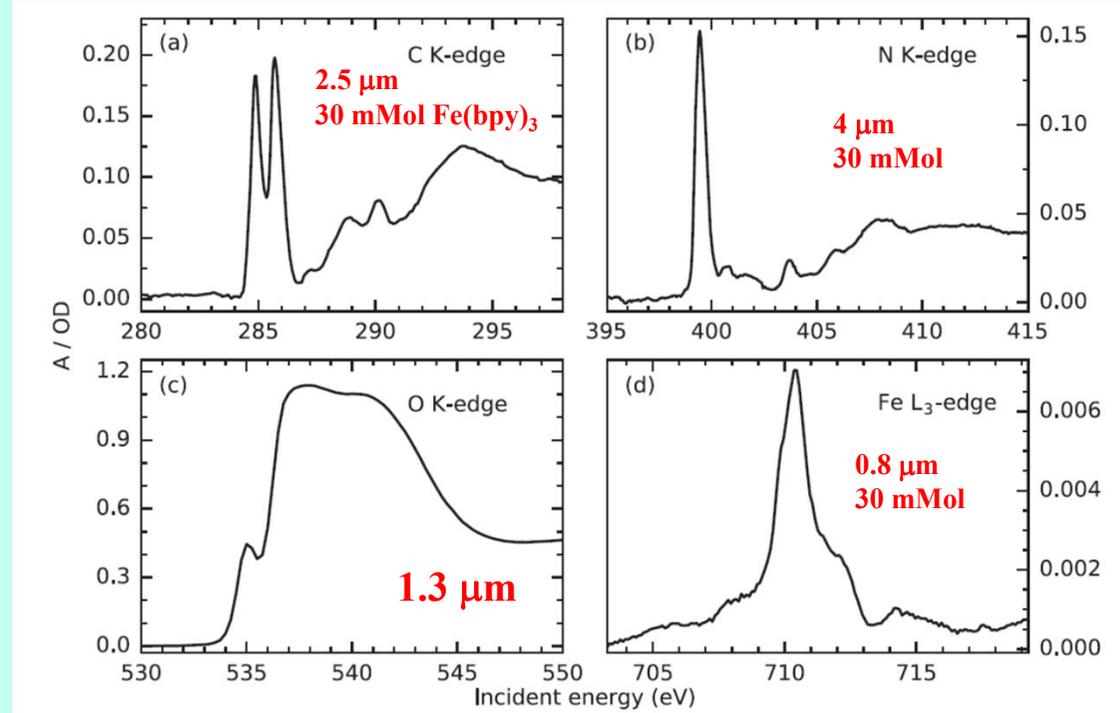
Aqueous Ferrioxalate



Flat liquid jets

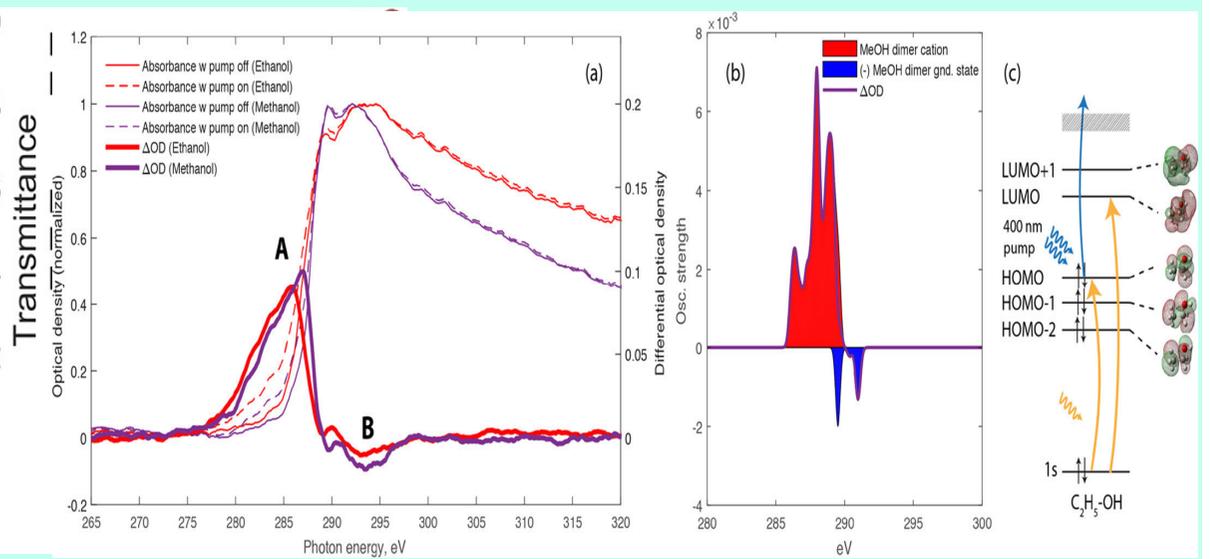
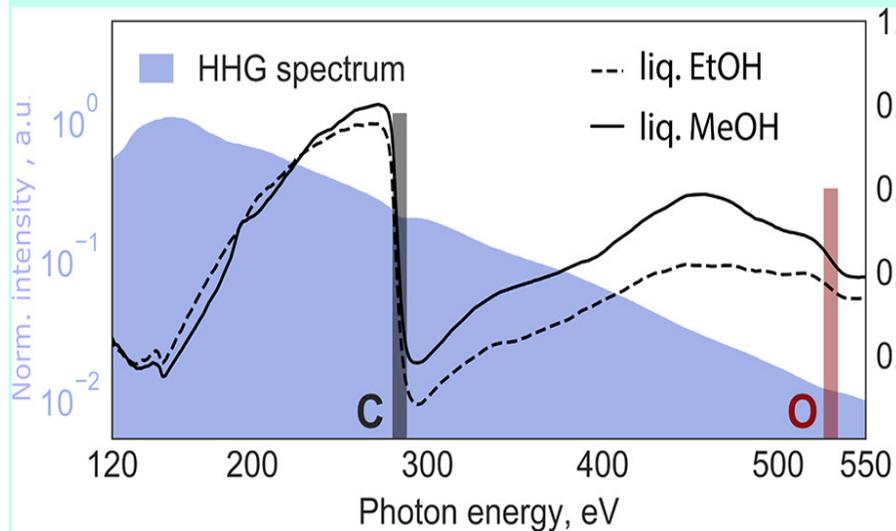


(Ekimova et al, Struct. Dyn. (2015))

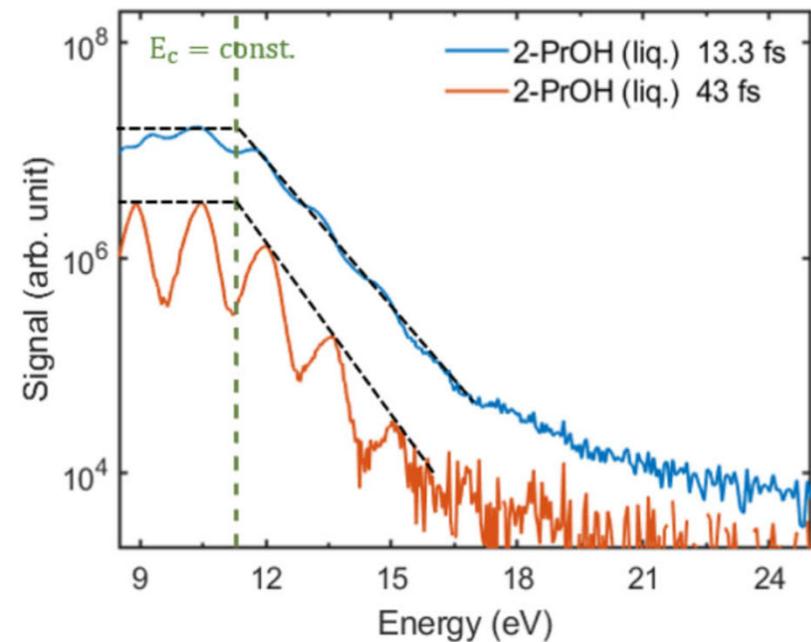
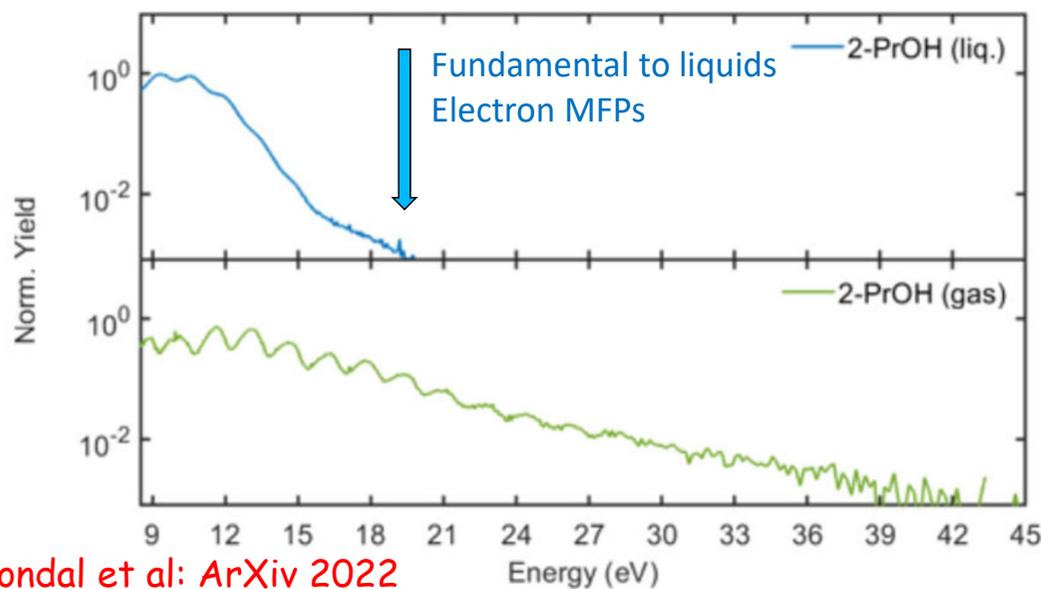
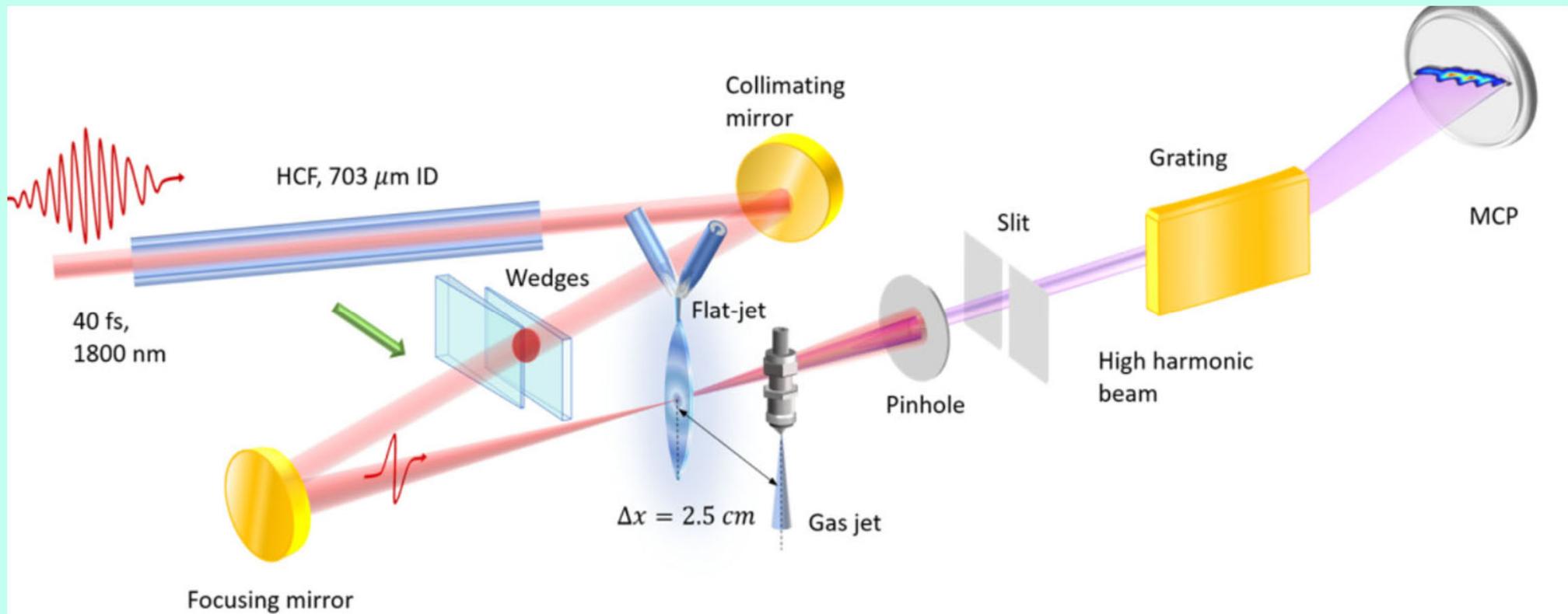


(Synchrotron studies: Fondell et al, Struct. Dyn. (2017))

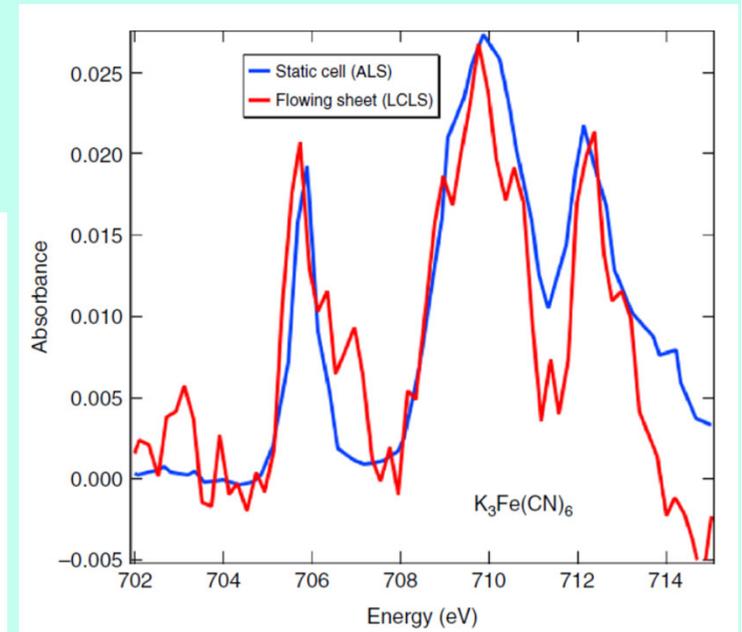
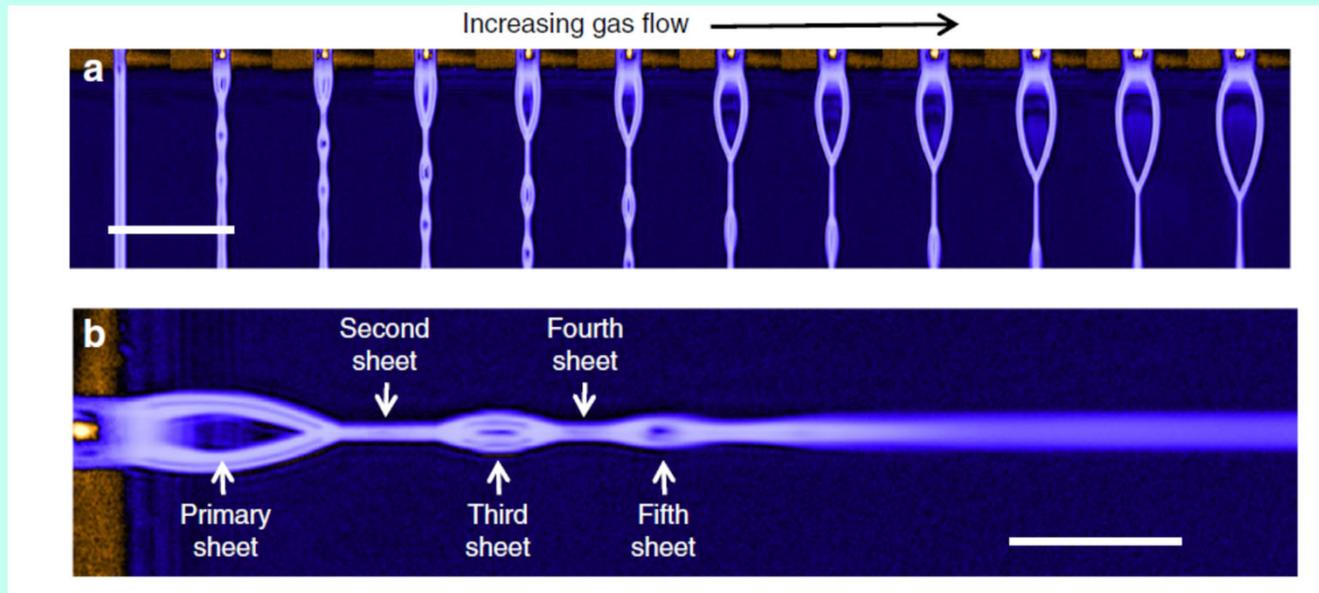
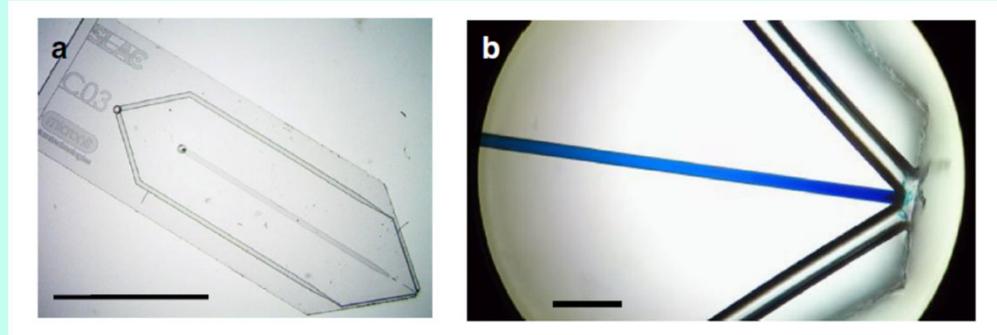
Tabletop HHG sources: Kleine et al: Phys. Chem. Lett. (2019) Smith et al: Phys. Chem. Lett. (2020)



HHG in flat liquid jets



Gas-compressed Flat liquid jets



De Ponte et al: Nature Comm. 2018

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